

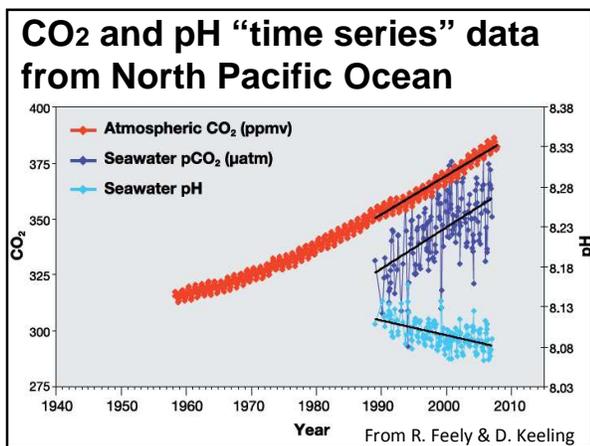
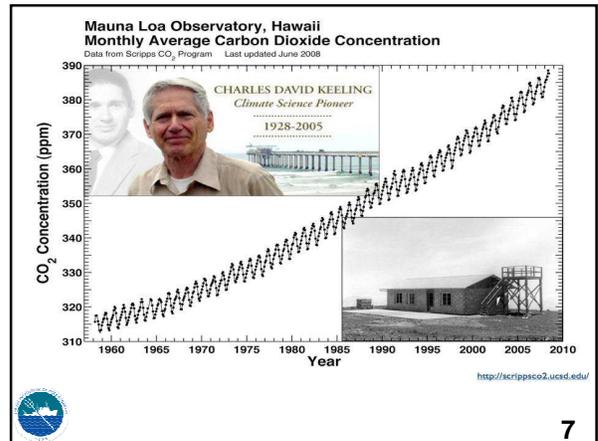
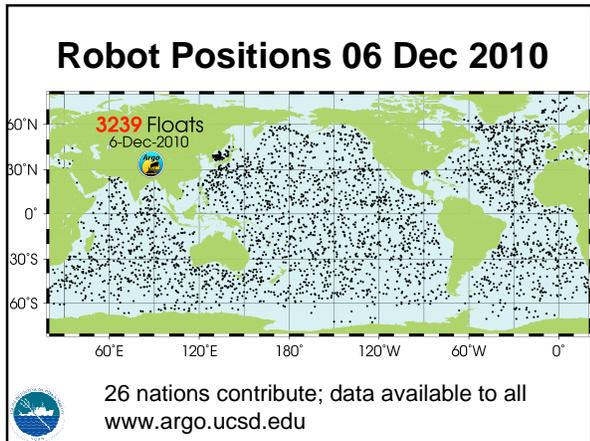
## SCRIPPS INSTITUTION OF OCEANOGRAPHY

*UC San Diego*

### Critical Needs in the Race to Observe Ocean Acidification

**Tony Haymet**

December 6, 2010 COP-16 Cancun



### Ocean Acidification

Since about 1850, the CO<sub>2</sub> chemistry of the oceans has been changing because of the uptake of anthropogenic CO<sub>2</sub> by the oceans.

- Decrease in pH of about 0.1 over the last two centuries; a projected decrease of 0.4 by 2100
- Today's ocean has undergone a 30% increase in acidity and a decrease in carbonate ion concentration of about 20%

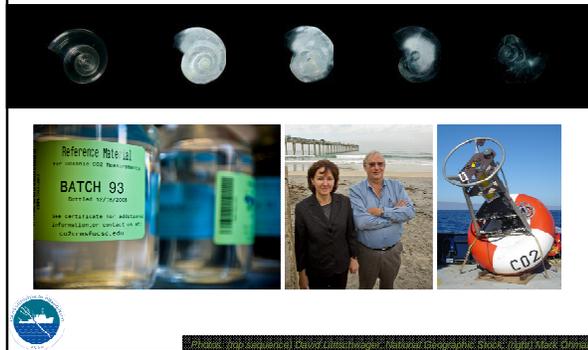
These changes in pH and carbonate chemistry may have profound impacts on many open ocean and coastal marine ecosystems.

Coral

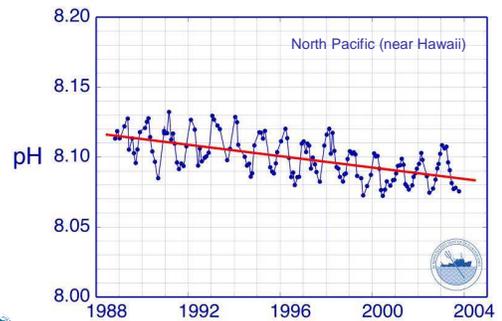
Calcareous Plankton

<http://www.biol.ksu.edu/~br/riose>

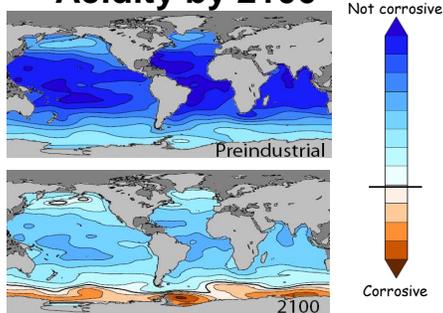
## Observing Ocean Acidification



## Declining Ocean pH

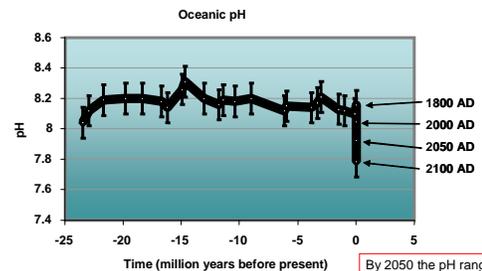


## Projected Changes in Ocean Acidity by 2100



Polar & some subpolar regions become corrosive (aragonite saturation < 1.0; pH < 7.75) as early as 2050 *Fabry et al. (2008)*

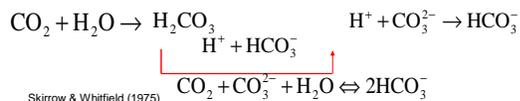
## At current rate we soon reach levels of ocean acidity (pH) not seen for 44 m yr



Past (from Pearson and Palmer, 2000) and predicted (from Turley et al., 2005) variability of marine pH. Future predictions are model derived values based on IPCC mean scenarios. The error bars indicate the likely seasonal scale variability in pH.

By 2050 the pH range will be discrete from the entire pre-industrial pH range. We will be in unknown territory.

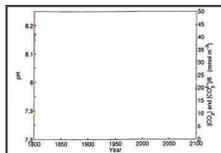
## CO<sub>2</sub>-induced seawater acidification: Simple chemistry



Skirrow & Whitfield (1975)

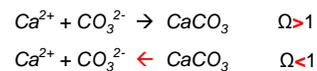
"The evolving chemistry of surface sea water under "Business as Usual"

Time	pCO <sub>2</sub>	Total CO <sub>2</sub>	pH	HCO <sub>3</sub> <sup>-</sup>	CO <sub>3</sub> <sup>2-</sup>	H <sub>2</sub> CO <sub>3</sub>
yr.	µatm	µmol kg <sup>-1</sup>		µmol kg <sup>-1</sup>	µmol kg <sup>-1</sup>	µmol kg <sup>-1</sup>
1800	280	2017	8.191	1789	217	10.5
1996	360	2067	8.101	1869	184	13.5
2020	440	2105	8.028	1928	161	16.5
2040	510	2131	7.972	1968	144	19.1
2060	600	2158	7.911	2008	128	22.5
2080	700	2182	7.851	2043	113	26.2
2100	850	2212	7.775	2083	97	31.8



Wolf-Gladrow, Riebesell, Burkhardt, Björn (1999)

## Calcification/carbonate dissolution



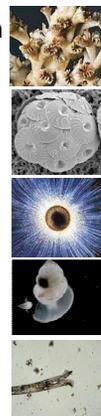
Saturation State

$$\Omega_{\text{phase}} = \frac{[\text{Ca}^{2+}][\text{CO}_3^{2-}]}{K_{\text{sp,phase}}}$$

$\Omega > 1 = \text{precipitation}$

$\Omega = 1 = \text{equilibrium}$

$\Omega < 1 = \text{dissolution}$

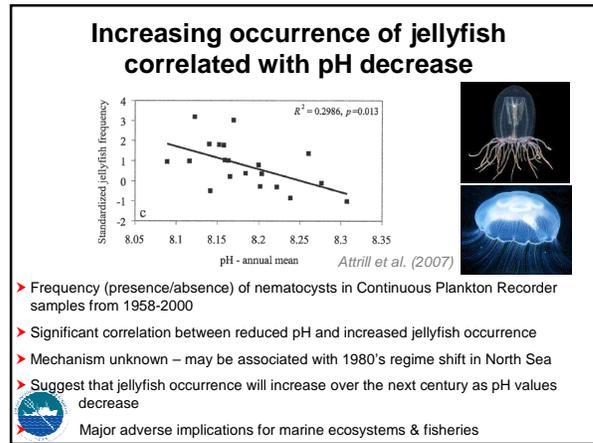
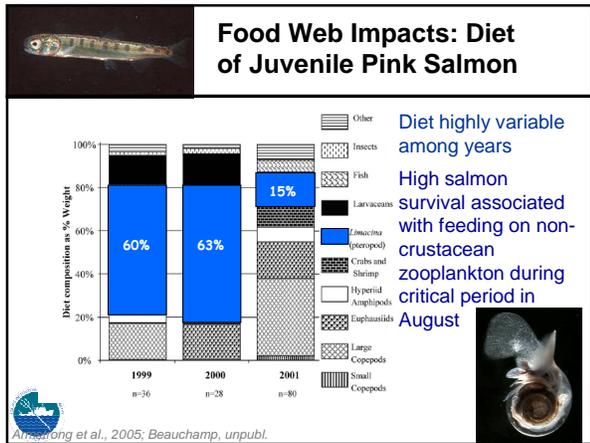




### Biological impact of increasing acidity: Shell dissolution in a live pteropod

NATURE 437, 29 Sept 2005  
J.C. Orr, ...Richard Matear,...

Shell from a live pteropod, *Clio pyramidata*, collected from the subarctic Pacific and kept in water undersaturated with respect to aragonite for 48 h. The whole shell (a) has superimposed white rectangles that indicate three magnified areas: the shell surface (b), which reveals etch pits from dissolution and resulting exposure of aragonitic rods; the prismatic layer (c), which has begun to peel back, increasing the surface area over which dissolution occurs; and the aperture region (d), which reveals advanced shell dissolution when compared to a typical *C. pyramidata* shell not exposed to undersaturated conditions (e).

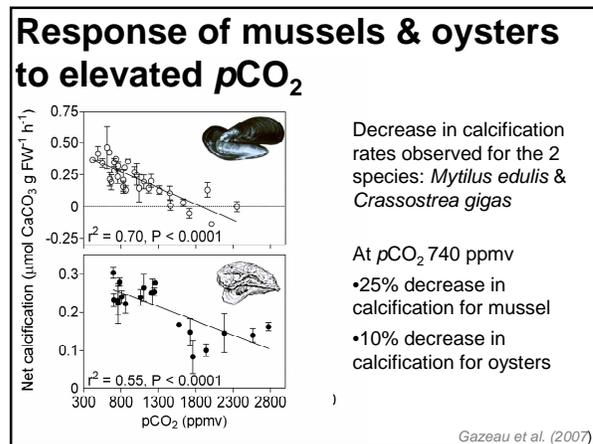


### Oyster larvae: Abnormal or no calcification in corrosive seawater

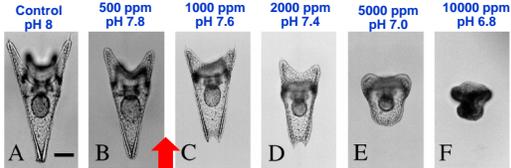
Normal seawater      Corrosive seawater

Oyster larvae did not form larval shells in corrosive seawater

Kurihara et al. (2007)



### Sea urchins: Impacts of increased CO<sub>2</sub> on development of larval stages



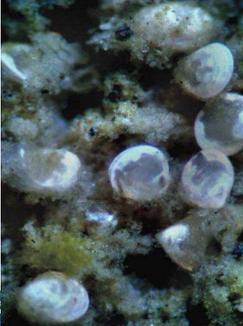
CO <sub>2</sub> - Seawater	Control pH 8	500 ppm pH 7.8	1000 ppm pH 7.6	2000 ppm pH 7.4	5000 ppm pH 7.0	10000 ppm pH 6.8
	A	B	C	D	E	F

pH of upwelled corrosive water off California (pH = 7.76)

- Malformed larval stages at high CO<sub>2</sub> and low pH
- Unable to calcify calcium carbonate skeletal rods at high CO<sub>2</sub> and low pH

Kurihara & Shirayama (2004)

### Clams – early life stages can also be sensitive to carbonate chemistry

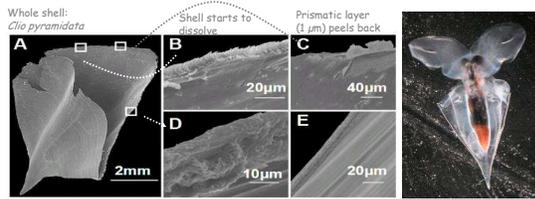


Hard shell clam *Mercenaria*

- Common in soft bottom habitats
- Used newly settled clams
- Size 0.3 mm
- Massive dissolution within 24 h in corrosive (i.e., undersaturated water); shell gone within 2 weeks

Green et al. (2004)

### Shells of living pteropods (plankton – food for salmon & other marine life) begin to dissolve at elevated CO<sub>2</sub> levels



Whole shell: *Gio pyramidata*

Shell starts to dissolve

Prismatic layer (1 μm) peels back

Edge of shell: advanced dissolution

Normal shell: Not exposed to corrosive water

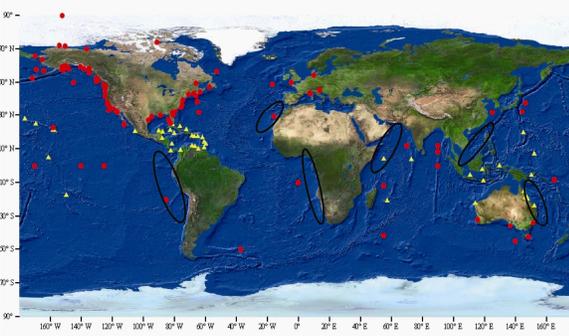
- Food source for marine predators including salmon, pollack
- Integral component of food webs in polar & subpolar regions

Orr et al. (2005)

### A Path Forward

- We know enough to act: reduce CO<sub>2</sub>
- To know precisely which commercial fisheries (& marine ecosystems) will be affected first...
- International measurement network: CO<sub>2</sub> as function of time and depth, available for all
  - Corals
  - high latitudes, coastal, & open ocean
- Communal facilities for live organism studies
- Integrated Modeling “from CO<sub>2</sub> to fish”

### An International Network

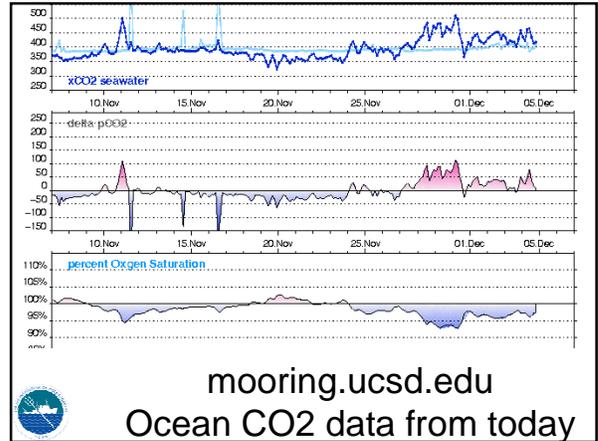
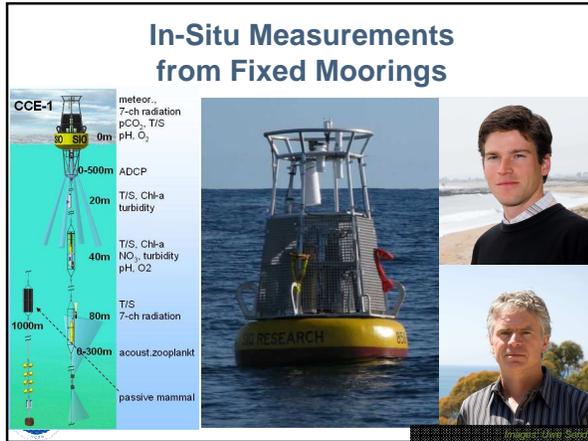
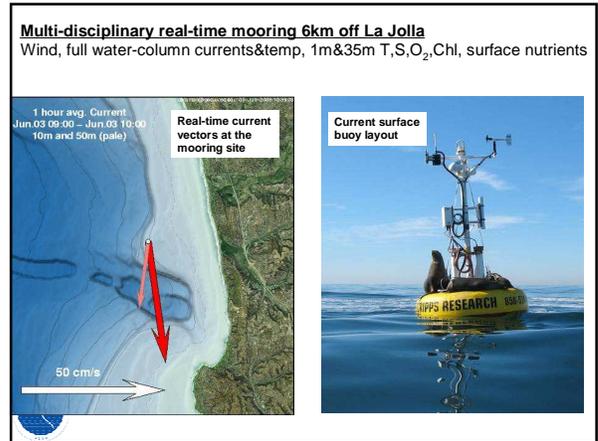
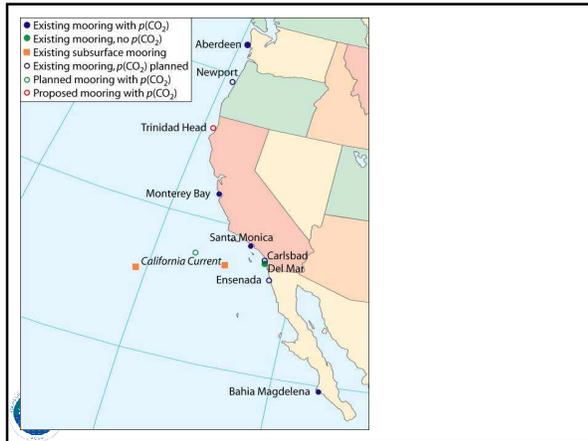


Map showing the international network of CO<sub>2</sub> observation stations across the globe, with stations marked by red dots and yellow triangles.

### ‘Baja to BC’ Test bed



- The groups already cooperate
- “Baja to BC” West Coast is one example which urgently needs ocean acidification observing system
- CO<sub>2</sub> observations will tell us where & when to look for biological effects
- Community experimental facility to test impacts on commercially & ecologically important species



**Critical Research Needs & Challenges**

- Standardized measurement protocols and data reporting guidelines (calcification rates, manipulation of seawater CO<sub>2</sub> system, dissolution, respiration)
- Autonomous systems for measurement of an additional parameters of the seawater CO<sub>2</sub> system
- Methods to investigate *in situ* response of organisms that are difficult to maintain in the lab (e.g., foraminifera, pteropods, squid)
- Methods to identify effects of chronic exposure to elevated pCO<sub>2</sub>

Address questions of species adaptation over timescales of decades

**Question: Why do we think we can make regional then global network?**

**Answer: Community has already built a network of 3000 robots for temperature and salinity over last 12 years: ARGO**

## Argo Floats



Davis and Roemmich with a float "robot"



### ROBOTS

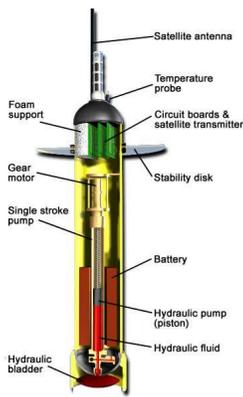
Depth

Temperature

Salinity

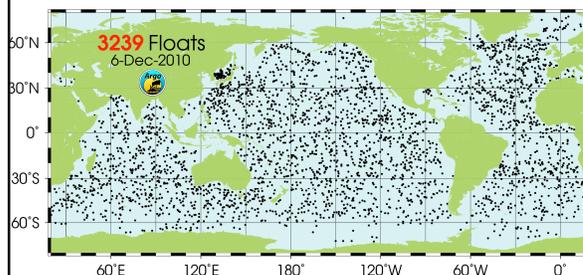
Need to be Extended to

Dissolved Gasses & biology



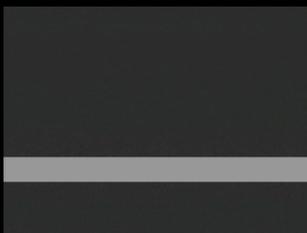
A profiling Argo float

## Robot Positions 06 Dec 2010



26 nations contribute; data available to all  
[www.argo.ucsd.edu](http://www.argo.ucsd.edu)

## Climate Ocean Observations



## A Lesson from Argo

- One proven route to an operational International network
- 1. Scientists do it first
- 2. Collect the data to "make the case"
- 3. Agencies and formal structures use that case to complete the network & make it "sustained"
- 4. "Light touch" open network structure survives the transition (much to the delight of scientists..)



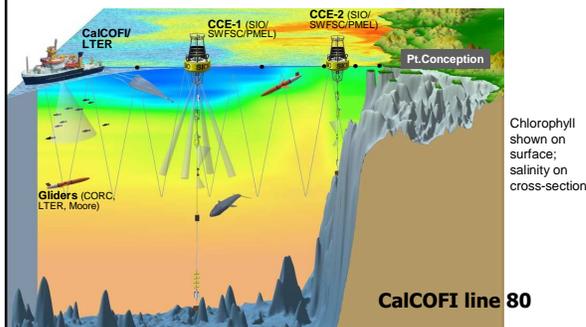
## One Path Forward

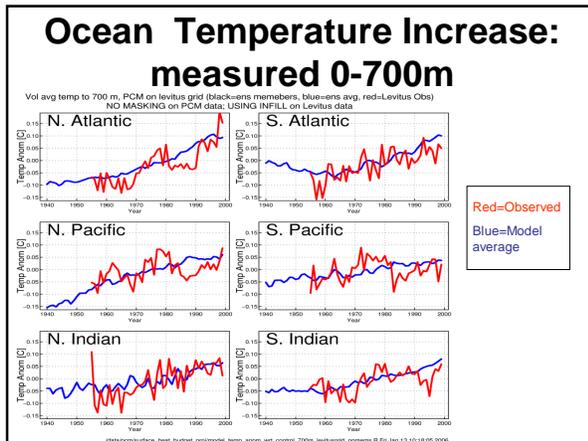
- International measurement network: CO<sub>2</sub> as function of time and depth, available for all
  - high latitudes, coastal, open ocean & corals
- Communal facilities for live organism studies – what is waiting for us? How long do we have?
- Integrated Modeling "from CO<sub>2</sub> to fish"
- Addressing Agency overload and "unfunded mandates"
- Economists, Social scientists, and marketing: avoiding the mistakes of CO<sub>2</sub> "heat"



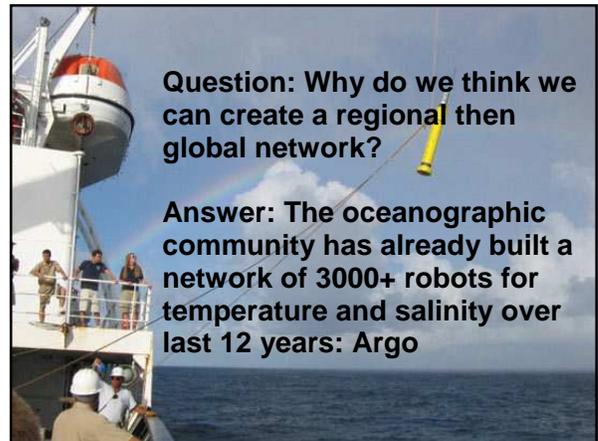
## The power of CCE1/2 comes from the context of other measurements

- Ships sample many variables and provide ground truth
- Gliders provide cross-shelf sampling with a few variables
- Moorings give full time sampling, wide range of variables





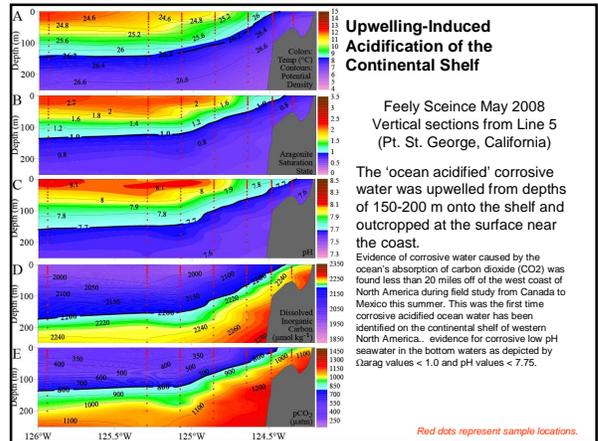
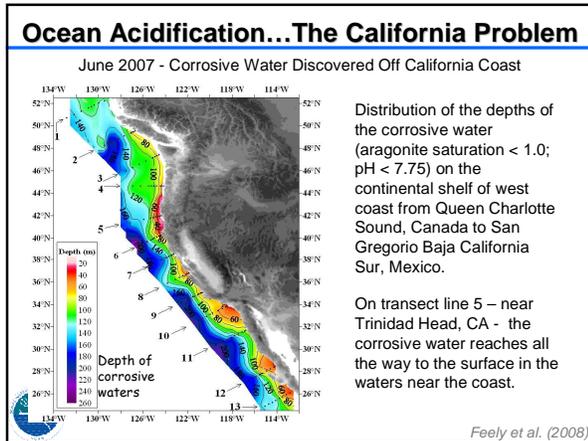
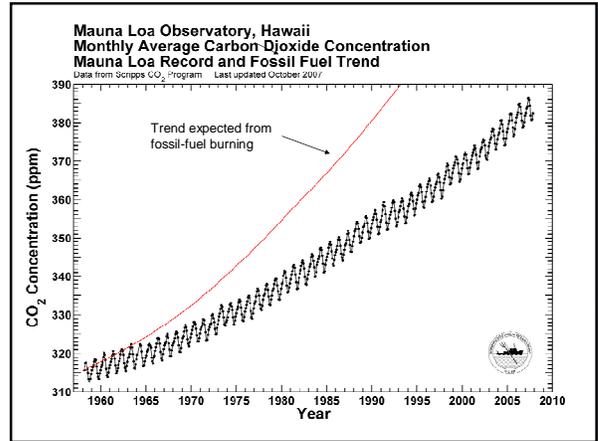
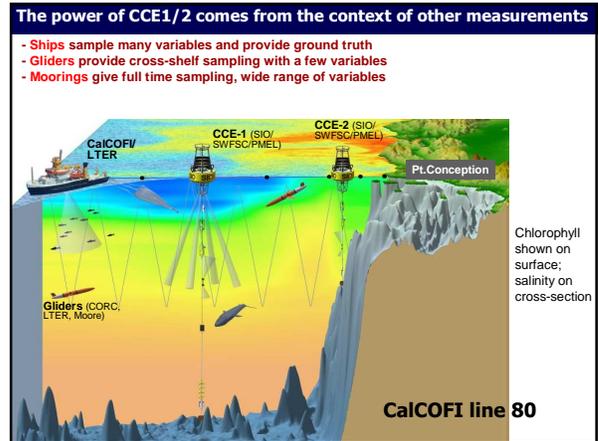
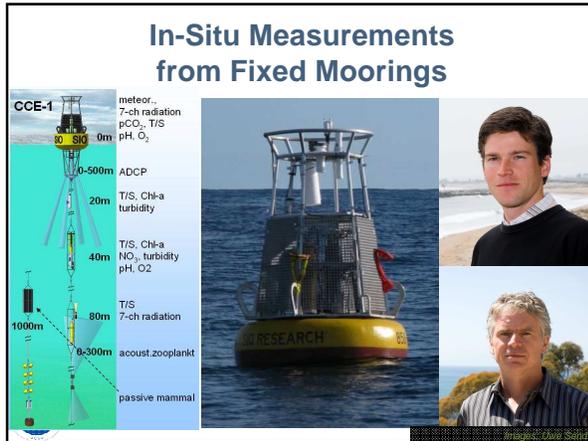
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## ‘Baja to BC’ Test bed

- The groups already cooperate
- “Baja to BC” West Coast is one example which urgently needs ocean acidification observing system
- CO<sub>2</sub> observations will tell us where & when to look for biological effects
- Community experimental facility to test impacts on commercially & ecologically important species



## Feely's Conclusions

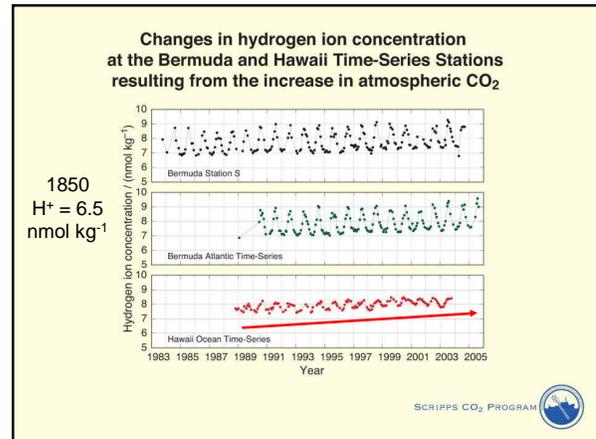
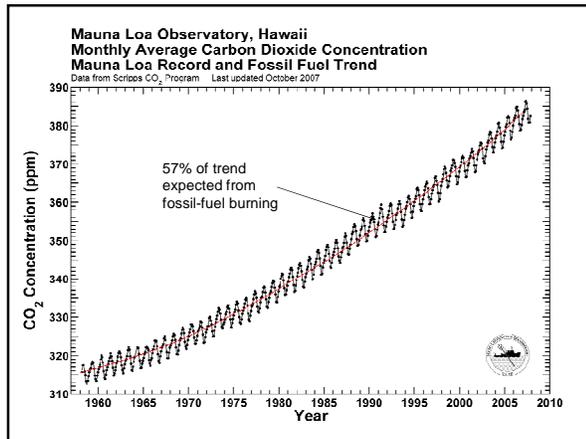
- Uptake of anthropogenic carbon dioxide in the North Pacific induces a  $1 \text{ m yr}^{-1}$  shoaling of the aragonite saturation horizon.
- Seasonal coastal upwelling of undersaturated water with respect to aragonite extends throughout the continental shelf region from southern Canada to northern Mexico.
- Possible responses of ecosystems are speculative but could involve changes in species composition & abundances - could affect food webs, biogeochemical cycles.



## Immediate Needs for the US West Coast



- Urgent need to develop US west coast ocean acidification observing system
- West coast network of  $\text{CO}_2$  observations will tell us where & when to look for biological effects
- Urgent need for experimental facility to test impacts on commercially and ecologically important species



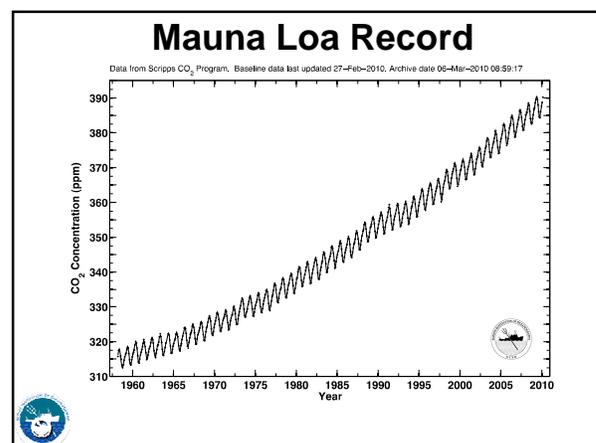
**Dave Keeling - Scripps 1957-2005**

- *Charles David Keeling (1928-2005)*

Image credit: Publication of the National Oceanic & Atmospheric Administration (NOAA), NOAA Central Library.  
 Photo Date: 1982 February; Photographer: Commander John Bortniak, NOAA Corps (ret.)

$\text{CO}_2$  Concentration (ppm)

1960 1964 1968 1972



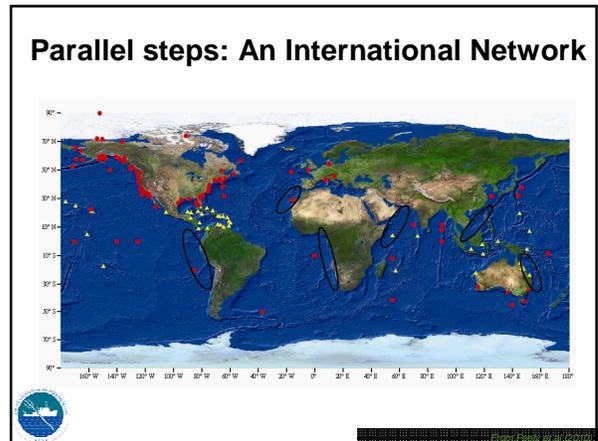
**Los Angeles Times** **A warning from the sea**  
 Oyster 'seeds' are dying as Pacific Coast waters grow warmer.  
 By Kenneth R. Weiss, Los Angeles Times Staff Writer  
 July 13, 2008

**Oyster farming**

Shellfish farms

- Number 1 aquacultured crop in the world
- \$3 billion industry
- Oysters are cultivated all along west coast (see map)
- Larval oysters dying - may be a result of warming temperature, bacteria, low oxygen and low pH
- West coast shellfish industry is at risk
- Without oysters & other filter feeding molluscs, estuaries may have more algal blooms & decreased water quality in future

source: Pacific Coast Shellfish Growers Assn.  
 Graphics reporting by KENNETH R. WEISS



**Ocean Acidification Becomes Warming's 'Evil Twin' at COP15**

UK warns on 'acidifying oceans'

the COP15 post

Acid oceans: Global warming's 'evil twin'

Den anden klimatrussel

Photo: Science Greenprints

